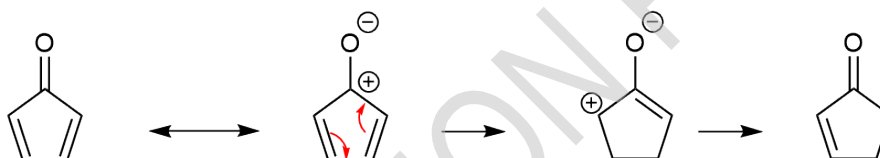


## Nazarov Reaction

5% of total							
Question	8.1	8.2	8.3	8.4	8.5	8.6	Total
Points	5	2	6	2	8	8	31
Score							

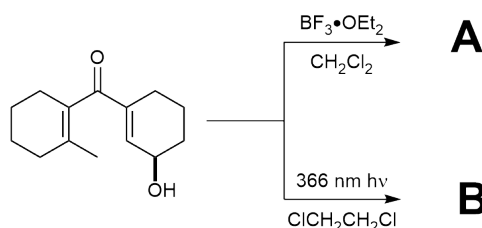
The Nazarov reaction is a frequently used reaction of divinyl ketones to give cyclopentenones. It proceeds either photochemically or via acid catalysis and is an electrocyclic reaction, followed by a proton transfer.



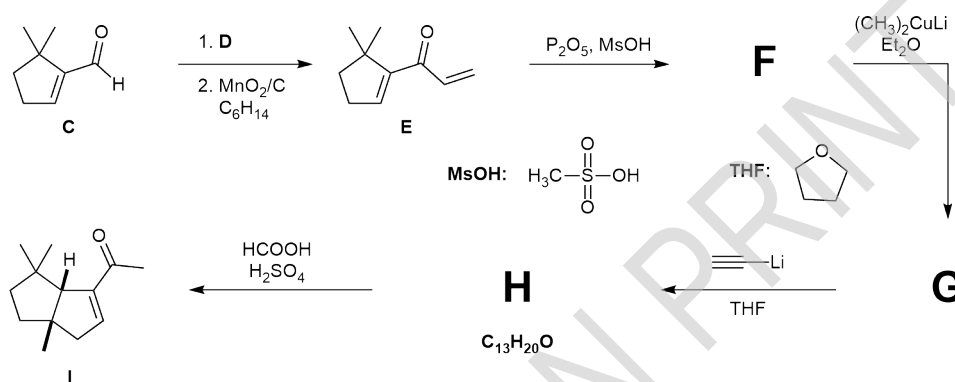
- 8.1** **Draw** the pi molecular orbitals to describe the Nazarov reaction. **Fill in** the electrons into the respective energy levels. **Mark** with an X the **i)** HOMO (highest occupied molecular orbital) and **ii)** LUMO (lowest unoccupied molecular orbital). For this exercise, you can consider the divinyl ketone as a pentadienyl cation with five p-orbitals. 5pt

- 8.2** From the pi molecular orbitals you derived in **Task 8.1**, **predict** under which conditions the Nazarov reaction of the divinyl ketone will proceed in a disrotatory or conrotatory fashion. In the **table on the answer sheet**, **mark** with an X the conditions under which the reaction is allowed. 2pt

- 8.3** The Nazarov reaction was used as key reaction in a synthesis of Farnesin. For both conditions below, **draw** one possible structure for each of **A** and **B**, including stereochemistry. Note that the products of both reactions show a signal at 6.70 - 6.73 ppm in the <sup>1</sup>H NMR. 6pt



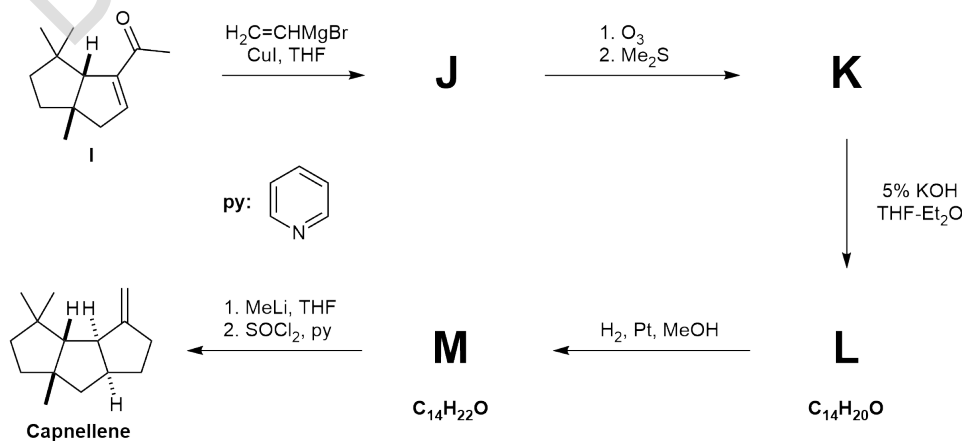
The synthesis of Capnellene commences with unsaturated aldehyde **C** shown below. Treatment with conditions **D**, followed by reaction with  $\text{MnO}_2$  supported on carbon gave divinyl ketone **E** shown below. Exposure to a mixture of  $\text{P}_2\text{O}_5$  and  $\text{MsOH}$  yielded **F**, which was elaborated via a sequence of reactions to the unsaturated ketone **I**.



**8.4** Choose the reagent(s) from the list on the answer sheet that would be suitable as **D**. 2pt

**8.5** Give the structures of intermediates **F**, **G**, and **H**, including their stereochemistry. 8pt

Enone **I** was then subjected to  $\text{H}_2\text{C}=\text{CHMgBr}$  and  $\text{CuI}$  in THF to give intermediate **J**, followed by ozonolysis to yield intermediate **K**, which shows a signal at 9.61 ppm in the  $^1\text{H}$  NMR. Treatment with 5%  $\text{KOH}$  in a mixture of THF and ether yielded intermediate **L**. Hydrogenation with a  $\text{Pt}$ -catalyst and under an atmosphere of  $\text{H}_2$  yielded **M**, which finally gave rise to Capnellene.



## Theory



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CHEMISTRY OLYMPIAD  
SWITZERLAND 2023

# Q8-3

English (Official)

**8.6** **Give** the structures of **J**, **K**, **L**, and **M**, including their stereochemistry.

8pt

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